

RESEARCH ARTICLE

A novel approach for qualitative biosensing of anti-leptospiral IgG antibodies in human sera using functionalised silver nanoparticles

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Abstract: Nanoparticles can interact with biological molecules, thus broadening the applications in the field of medicine. Surface plasmon resonance of nanoparticles in ultraviolet-visible (UV-visible) region can be applied to identify nanoparticle-protein interactions. This property is used to detect antibodies against leptospirosis, which is an emerging infectious zoonosis in many developing countries and endemic in Sri Lanka. A rapid screening test to detect anti-leptospiral antibodies in human sera by analysing antigen-antibody interactions using UV-visible spectroscopy is reported. The protocol uses hydroxylamine hydrochloride stabilised silver nanoparticles followed by functionalisation with rLigA (recombinant leptospiral immunoglobulin-like protein A) antigens. rLigA-functionalised silver nanoparticles were characterised by UV-visible spectroscopy, dynamic light scattering (DLS), Fourier transform infrared (FT-IR) spectroscopy and scanning electron microscopy (SEM). A monodispersed solution of LigA-functionalised silver nanoparticles was prepared at 25 °C and pH 7.0 with optimum LigA concentration of 1.5 µg mL⁻¹. This localised surface plasmon resonance (LSPR) biosensor shows a positive response to anti-leptospiral IgG antibodies at optimum conditions in dilutions lower than 1:25 (SD = ± 0.047; 95 % CI = 1.100–1.126). The dilutions, 1:300 to 1:50 of IgG antibodies maintained no response compared to the nanoparticle conjugate. Furthermore, the non-responsive UV-visible spectra of nanoparticle conjugate with anti-leptospiral antibody negative serum samples confirmed that the conjugate was specific to anti-leptospiral antibodies (SD = ± 0.012; 95 % CI = 1.177–1.183). This novel technique offers substantial improvement in terms of screening time, which does not take more than 30

minutes, compared to conventional methods such as ELISA consumes 2–3 hours.

Keywords: Anti-leptospiral antibodies, functionalisation, leptospirosis, localised surface plasmon resonance, nanoparticles, UV-visible spectroscopy.

INTRODUCTION

In the 20th century, the fields of medicine and pharmaceuticals have been revolutionised with the advancement of nanotechnology, introducing nanomedicine as the latest innovation in the field of science (Watkins *et al.*, 2015; Manatunga *et al.*, 2018; Patra *et al.*, 2018; Karunaratne *et al.*, 2019). Nanoparticles can interact with biological molecules which are in nano-range, thus broadening the applications in the field of medicine (Wang *et al.*, 2009). Metallic nanoparticles have unique morphologies, surface plasmon characteristics, quantum confinement, short-range ordering, physicochemical properties and large surface energies (Blackman, 2008). The characteristics of metallic nanoparticles depend on their size, shape, crystallinity, structure and dielectric constant of the medium, leading to many studies and applications. These parameters can be changed or controlled to achieve desired properties of nanoparticles (Blackman, 2008).

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Metallic nanoparticles can be modified using various chemical functional groups which allow them to conjugate with nucleic acids, ligands, proteins and drugs of interest, enabling detection and manipulation and providing good stability. This opens a variety of potential applications in biotechnology and medicine such as magnetic separation (Holla *et al.*, 2015), biosensing (George *et al.*, 2018), targeted drug delivery (Kumari *et al.*, 2016), vehicles for gene and drug delivery (Maiyo & Singh, 2017), drug encapsulation (Wais *et al.*, 2016) and diagnostic imaging (Soloviev, 2004; Nikalje, 2015; Karunaratne *et al.*, 2019).

Metallic nanoparticles, especially silver and gold nanoparticles, possess excellent optical properties such as localised surface plasmon resonance (LSPR), surface-enhanced Raman scattering (SERS), fluorescence, etc. Silver nanoparticles possess the highest capacity of plasmon excitation among metallic nanoparticles, which show plasmon resonance in the visible range (Priyanka *et al.*, 2013). LSPR wavelength and width depend on particle size, shape, medium and surface coating. The influences of above features, such as the binding of specific analytes on nanoparticle surface which alters the dielectric environment of the nanoparticle, can be identified by the colour change of solution or LSPR peak shift of absorption band (Hong *et al.*, 2012). Peak shift can be measured using UV-visible spectrophotometer.

Leptospirosis is recognised as the most widespread bacterial zoonosis (Cosson *et al.*, 2014). It is a major public health problem in many developing countries and endemic in India, Thailand, Indonesia and Sri Lanka. The disease occurs worldwide throughout the year with a seasonal distribution and causes morbidity and mortality in regions with high rainfall and high temperature, especially tropical and subtropical areas (Pappas *et al.*, 2008). Current statistics assert that the annual worldwide cases of leptospirosis are above one million (Cosson *et al.*, 2014).

Leptospirosis, caused by pathogenic *Leptospira* spp., is known as an acute febrile illness with a wide clinical spectrum which ranges from a mild, influenza-like illness to a critical infection such as renal failure, hepatic failure, pulmonary distress, jaundice (icterus) and also death, referred to as the classical Weil's disease (Fraga *et al.*, 2014). Clinical diagnosis of leptospirosis is rather difficult as the symptoms are nonspecific. Diseases such as influenza, dengue, Hantavirus infection, yellow fever, malaria and hepatitis should be considered in the differential diagnosis and laboratory tests should be carried out to confirm the infection (Toyokawa *et al.*,

2011). The diagnosis can be performed by detecting antibodies or nucleic acids, by culturing or by verifying the presence of bacteria in tissues (Budihal & Perwez, 2014; Chirathaworn *et al.*, 2014; Khaki, 2016).

Diagnosis of most of the cases are done through serology. ELISA is one such serological test, which can detect IgM and IgG antibodies in human sera, indicating current/ recent infection and exposure to pathogen/ past infection, respectively (Khaki, 2016). ELISA detects the presence of proteins by colour change, which can be identified by the naked eye. However, some ELISA tests are low in specificity and positive results for the disease may be observed due to the presence of other antibodies as well (WHO, 2003). Therefore, PCR (polymerase chain reaction) or MAT (microscopic agglutination test) can be used along with ELISA to confirm the diagnosis (Budihal & Perwez, 2014). Furthermore, ELISA requires several washing steps and incubation steps, which are time consuming, extending the analysis up to 3 to 5 hours. Thus, an alternative method to overcome the limitations of this diagnostic technique is required.

Nanoparticle-based immunoassays contribute to diagnostic systems for rapid and efficient identification of many diseases due to the unique properties of nanoparticles. In this study, a rapid screening test (detection of IgG does not confirm acute disease but past infection) to detect anti-leptospirosis antibodies by analysing antigen-antibody interactions using silver nanoparticles was developed, which can be used for population-based screening purposes. UV-visible spectroscopy was used as the principal analysis method and dynamic light scattering (DLS), Fourier transform-infrared (FT-IR) spectroscopy and scanning electron microscopy (SEM) were used for the characterisation. Different environmental parameters that might affect the conjugation between silver nanoparticles and antigens, such as temperature, pH and protein concentration, were investigated in advance. Furthermore, antigen-coated silver nanoparticles were tested with anti-leptospirosis antibodies to obtain the detection limit and sensitivity of the developed system. The formation of aggregates upon the addition of anti-leptospirosis antibodies was identified by the decrement of UV-visible absorption peak and the increment of hydrodynamic diameter. Here, the antibody capturing by antigens is critical and depends on the immobilisation strategy. Conjugation between the nanoparticle and antigen should be stable and the antigens should be in biologically active molecular conformation with minimum changes by means of high accessibility for capturing antibodies from the solution.

METHODOLOGY

Reagents

Silver nitrate, hydroxylamine hydrochloride, sodium hydroxide, sodium chloride, potassium bromide and bovine serum albumin (BSA) were purchased from Sigma-Aldrich. LigA recombinant protein fused with GST (glutathione-S-transferase) was supplied by Dr Nobuo Koizumi, National Institute of Infectious Diseases, Japan. Both anti-leptospiral IgG antibody positive and negative human serum samples were provided by the Department of Microbiology, Faculty of Medicine, University of Peradeniya, Sri Lanka. Ethical clearance was not sought as the said serum samples were opted-in by patients for further use of diagnostic developments.

Synthesis and characterisation of Ag NPs

All laboratory glassware, magnetic stir bars and other equipment used in the experiment, were thoroughly washed with soap, in aqua regia (HCl/HNO₃ 3:1, v/v) and with distilled water, respectively. Those were oven dried before usage to avoid undesirable nucleation during the analysis. Deionised (DI) water was used in all the experiments when necessary. Silver nanoparticles were synthesised according to Leopold and Lendl method (Leopold & Lendl, 2003) using silver nitrate as the precursor and hydroxylamine hydrochloride containing sodium hydroxide as the reducing agent. Briefly, 90.00 mL of $1.67 \times 10^{-3} \text{ mol dm}^{-3}$ hydroxylamine hydrochloride aqueous solution containing $3.33 \times 10^{-3} \text{ mol dm}^{-3}$ sodium hydroxide was initially prepared at room temperature (25 °C), by maintaining the pH at 6.0, 7.0, 8.0 and 9.0. Then, 10.00 mL of $1.00 \times 10^{-2} \text{ mol dm}^{-3}$ silver nitrate solution was added drop-wise to each solution under vigorous stirring on a magnetic stirrer at 25 °C. Stirring was continued for additional 10 min. The above procedure was repeated at 37 °C and the optimum temperature for the synthesis of silver nanoparticles was identified according to UV-visible spectroscopy (Shimadzu UV 1800 Series).

The characterisation of silver nanoparticles was done using particle size analysis (CILAS- Nano DS), scanning electron microscopy (ZEISSEVOLS15) and UV-visible spectroscopy. The nanoparticles were kept at 4 °C for further use. Further investigations were carried out at optimum temperature.

Optimisation of antigen concentration and environmental conditions for silver nanoparticle functionalisation

Initial concentration of recombinant LigA (leptospiral immunoglobulin-like protein A) was measured using the Nanodrop spectrophotometer (Thermo Scientific 2000). Different volumes of LigA (sample size, n = 6) were added to Kahn tubes containing 4.00 mL of silver nanoparticles of pH 6.0, 7.0, 8.0 and 9.0 to obtain a concentration series from 0.5–2.5 $\mu\text{g mL}^{-1}$ (namely, 0.5, 1.0, 1.5, 2.0, and 2.5 $\mu\text{g mL}^{-1}$) of the antigen at each pH level. The prepared samples were slightly vortexed and incubated at room temperature (25 °C) for 1 h. Each sample was divided into two parts and one part was used to measure the absorbance using UV-visible spectrophotometer, while the other part was used for DLS measurement. Optimum pH level and LigA concentration were identified using the above UV-visible spectra and those results were further analysed using DLS.

DLS was used to screen the hydrodynamic diameter of particles. For this, 0.4 μg of 10 % (mass ratio) NaCl was added to each tube to verify the stability of nanoparticle-protein conjugates in saline environment (Genevieve *et al.*, 2007). DLS measurements were taken before and after the salt addition for confirmation of optimum LigA concentration and optimum pH level, which were initially identified by UV-visible spectra. Further investigations were carried out at the above environmental conditions.

Further characterisation of samples was done using SEM and FT-IR (fourier transform-infrared) spectroscopy. In order to carryout SEM analysis, the samples were sonicated for 20 min in an ultrasonic bath to obtain an evenly dispersed solution by removing the aggregates. The samples were then placed on the sample stub and kept in an evaporation chamber for nearly 12 h. Carbon tape was used afterwards to adhesively bind the samples to the stub and the samples were gold sputtered before the analysis. For FT-IR spectroscopy, the samples were freeze-dried and diluted with potassium bromide in 1:100 ratio. SEM produces the images of a sample by scanning the surface with a focused electron beam, while FT-IR spectroscopy is used to identify the changes in a protein structure. Proteins contain amide linkages between amino acid residues, which possess well known peaks in the infrared region. FT-IR spectrum of a complex protein is composed of many overlapping bands and important information can be hidden in those undistinguishable, broad absorption bands (Barth, 2007). BSA was used as a secondary stabiliser for the

functionalised nanoparticles, thus, 0.25 % BSA was added to the solution to achieve the required stability level. The nanoparticles can be stored at 4 °C for 4 wk or at room temperature for 2 wk for further use.

Analysis of the sensitivity of LigA-functionalised silver nanoparticles

For the determination of the presence or absence of IgG antibodies in human serum samples, *Leptospira* IgG ELISA and IFA were carried out to confirm the presence of anti-leptospiral IgG antibodies. Different volumes of human sera containing anti-leptospiral antibodies (n = 50) were added to Kahn tubes containing 4.00 mL of LigA-functionalised silver nanoparticles (the diluent) to obtain a dilution series from 1:300 to 1:5 (namely, 1:300, 1:200, 1:150, 1:100, 1:75, 1:50, 1:25, 1:20, 1:15, 1:10, 1:5) of the antibody. The solutions were allowed to incubate for 30 min at room temperature. The aggregation process was analysed by UV-visible spectroscopy and SEM.

Evaluation of the specificity of LigA-functionalised silver nanoparticles

In order to evaluate the specificity of conjugated nanoparticles, a control group of LigA-functionalised nanoparticles was tested with anti-leptospiral IgG antibody negative (previously tested with *Leptospira* IgG ELISA and IFA) serum samples (n = 20). Different volumes of serum were added to Kahn tubes containing 4.00 mL of LigA-functionalised silver nanoparticles to obtain a dilution series from 1:300 to 1:5 (namely, 1:300, 1:200, 1:150, 1:100, 1:75, 1:50, 1:25, 1:20, 1:15, 1:10, 1:5). Results were analysed by UV-visible spectroscopy.

RESULTS AND DISCUSSION

Synthesis of silver nanoparticles

Figure 1 represents the UV-visible absorption spectra of silver nanoparticles at 25 °C and 37 °C.

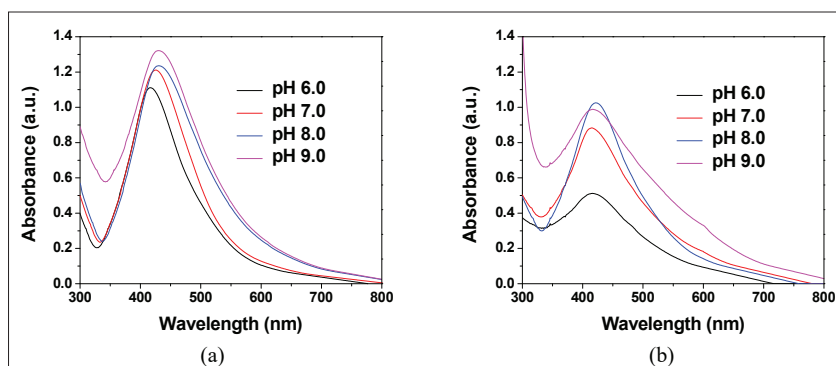


Figure 1: UV-visible absorption spectra of silver nanoparticles prepared at (a) 25 °C and (b) 37 °C at different pH levels.

Temperature has a great influence on the reactive kinetics of molecules. Herein, 25 °C and 37 °C were used to investigate the optimum temperature for the synthesis of silver nanoparticles. At 37 °C, relatively broader absorption bands with decrement of absorbance were observed at different pH levels due to aggregation of particles at high temperature, which is shown in Figure 1 (b). Therefore, 25 °C was ideal for the functionalisation process as narrow absorption bands indicated the high stability of nanoparticles. The differences in absorption bands may be due to the increase in kinetic energy with temperature. Particles move faster at high temperatures and make more collisions. As a result, particles tend to aggregate, rather than forming a homogeneous solution.

Functionalisation of silver nanoparticles with LigA antigens

In order to attach proteins to the nanoparticle surface, proteins were added to the nanoparticle solution drop wise under vigorous stirring for nearly 30 min at room temperature. Upon the addition of LigA, the proteins get attached to the nanoparticle surface due to the interaction with hydroxylamine chloride ions. The optimum concentration of LigA to cover a full monolayer is known as critical concentration.

Below the critical concentration, nanoparticles are under-saturated, i.e. partially covered with antigens,

whereas beyond the critical concentration, excess antigens are present in the solution and they can be loosely attached to the nanoparticle surface. At under-saturation level, many of the anti-leptospiral antibodies may not interact specifically with nanoparticles. At over-saturation level, antibodies get loosely attached to nanoparticles due to the presence of excess LigA. Therefore, it is crucial to obtain the saturation level of nanoparticles for further analysis using anti-leptospiral antibodies. The process of protein adsorption onto nanoparticle surface depends on several factors such as temperature and pH of the medium, concentration, isoelectric point of protein and ionic strength (Saha *et al.*, 2014; Taghipour *et al.*, 2018). Nevertheless, it is not possible to study all the above parameters simultaneously for the optimisation process.

The pH level of the medium affects the hydrogen bonds and overall charge of the protein. Exceedingly high or low pH cause changes in molecular conformation and the bioactivity. It is well-known that a pH equal or slightly above the isoelectric point (pI) of the protein is the optimum pH for adsorption of proteins to nanoparticle

surface (Taghipour *et al.*, 2018). For pH optimisation, the pH range was selected as 6.0 – 9.0 as the pI of LigA protein is about 6.5. Different amounts of LigA were added to pH adjusted silver nanoparticles to obtain a concentration series of LigA from 0.5 – 2.5 $\mu\text{g mL}^{-1}$ at each pH level. Figure 2 represents the surface plasmon resonance spectra of LigA functionalised silver nanoparticles at different pH levels.

The results are in agreement with the above mentioned statement as 7.0 was the optimal pH for protein adsorption, which was slightly above the respective pI. It has shown that the silver colloids of pH 6.0, 8.0 and 9.0 had unstable spectra with broad peaks, confirming particle aggregation compared to the silver colloids of pH 7.0. According to SPR spectra of pH 7.0, the highest absorption was obtained when the LigA concentration was 1.5 $\mu\text{g mL}^{-1}$ and that shows the most similar spectrum to that of nanoparticles as well. Therefore, 1.5 $\mu\text{g mL}^{-1}$ was taken as the optimal LigA concentration for the functionalisation. However, these results were further analysed using DLS.

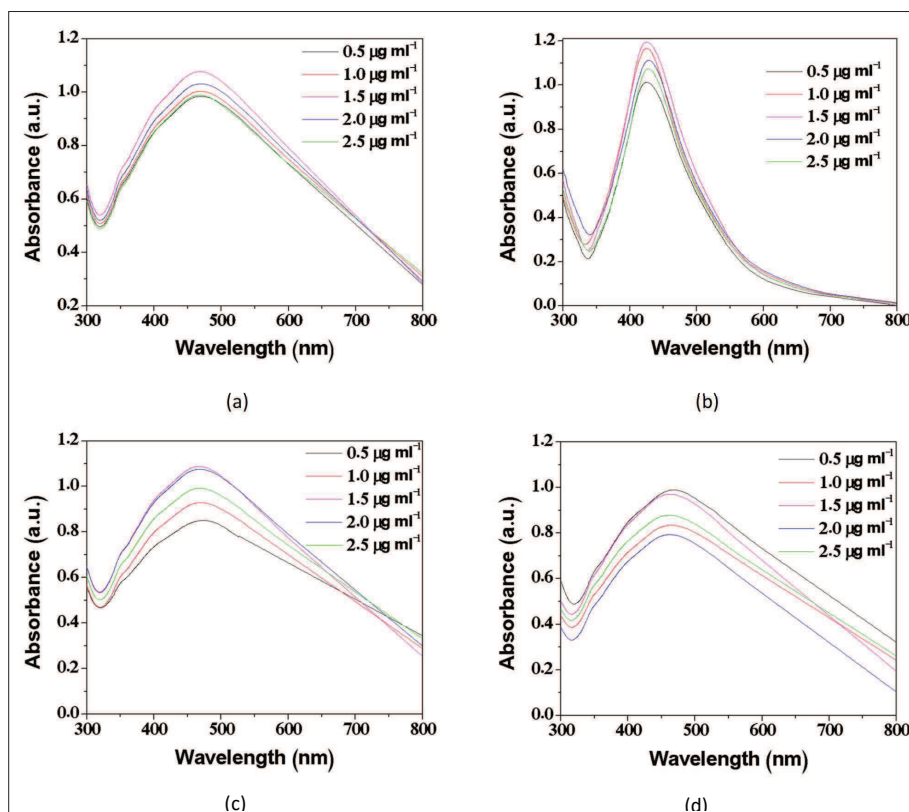


Figure 2: UV-visible absorption spectra of functionalised nanoparticles at pH levels of (a) 6.0, (b) 7.0, (c) 8.0 and (d) 9.0

If a monolayer of LigA was coated on nanoparticle surface, Na^+ ions in saline solution would not attach to hydroxylamine chloride ions, thus, it will not affect the stability of the conjugate. If an insufficient amount of LigA was present, most of the nanoparticles would still be covered with hydroxylamine chloride ions. As a result, the negative charge of nanoparticles would be screened by counter Na^+ ions, causing agglomeration. Using this experiment, it is easy to determine the critical concentration by gradually increasing the LigA concentration from a low value where flocculation in NaCl solution occurs, to a high value where flocculation is prevented.

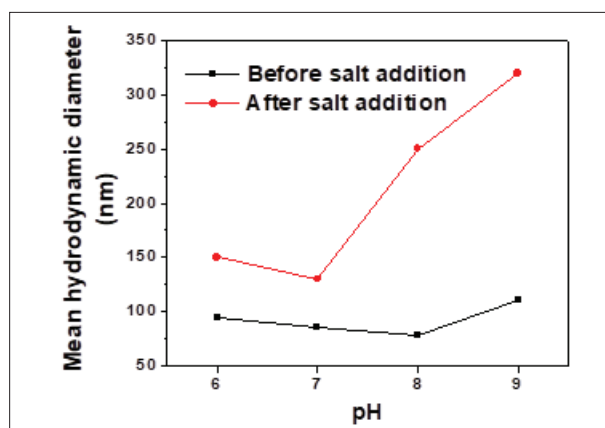


Figure 3: Mean hydrodynamic diameter of Ag NP-LigA conjugates (LigA concentration is $1.5 \mu\text{g mL}^{-1}$) as a function of pH, before and after addition of NaCl

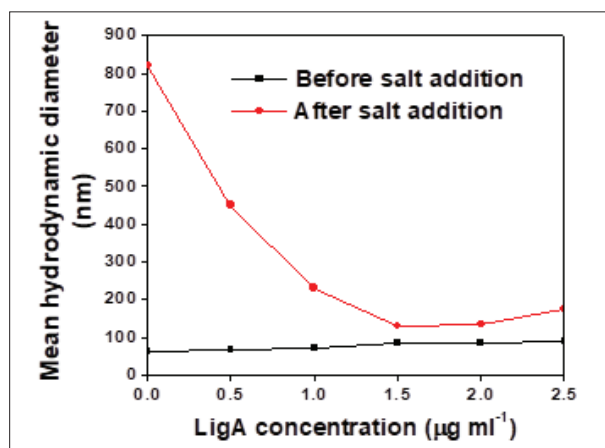


Figure 4: Mean hydrodynamic diameter of Ag NP-LigA conjugates (pH of 7.0) as a function of LigA concentration, before and after addition of NaCl

Figure 3 shows the adsorption curve plotted as mean D_H (hydrodynamic diameter) versus pH with a fixed LigA concentration of $1.5 \mu\text{g mL}^{-1}$, before and after the addition of NaCl. Correspondingly, Figure 4 shows mean D_H of LigA functionalised nanoparticles at pH of 7.0 as a function of LigA concentration, followed by the addition of NaCl.

The D_H of plain silver nanoparticles was measured as 60 nm. D_H was enhanced upto around 90 nm due to the adsorption of antigens depending on LigA concentration. At pH 7.0, only $1.5 \mu\text{g}$ of LigA was required per 1 mL of nanoparticles to form a monolayer of antigen on nanoparticle surface. It is likely that the molecular conformation of LigA can form several contacts with the nanoparticle surface so that it rapidly reaches saturation with a minimal amount of antigen. The stability level was also maintained at high LigA concentrations. These results of antigen adsorption and conjugate stabilisation were further evaluated by the DLS aggregation curve taken after the addition of NaCl. Flocculation had taken place in solutions with a lower concentration of LigA (0.5 and $1.0 \mu\text{g mL}^{-1}$) due to the absence of a sufficient amount of LigA to stabilise the nanoparticles. Conversely, flocculation was prevented by the presence of relatively high quantities of LigA molecules, as they stabilise the nanoparticles by forming a surface coating.

D_H of conjugated nanoparticles changes according to the pH of medium by 10 nm to 30 nm with respect to unconjugated nanoparticles. At lower pH level (6.0), the antigen has a net positive charge and the antigen itself destabilised the nanoparticle, inducing aggregation. As a result, D_H is increased by 34 nm. On the contrary, at pH level of 8.0, D_H was increased by only 18 nm, where the antigen was negatively charged and a full monolayer of antigen was not adsorbed onto the negatively charged hydroxylamine chloride capped nanoparticle. However, at pH 9.0, D_H has increased by 50 nm, as the extreme negative charge of LigA induced the aggregation of particles. NaCl was then added to investigate the antigen adsorption and subsequent stabilisation in solutions of high ionic strength. According to Figure 4, the D_H of antigen-nanoparticle conjugate has increased at extreme pH levels after the addition of NaCl, confirming that not enough antigens got adsorbed onto the nanoparticle surface to stabilise the nanoparticles. The antigen-nanoparticle conjugate was least affected by the electrolytes at pH 7.0, stabilising the nanoparticles by forming a full monolayer of antigens.

Thus, the DLS data were consistent with the UV-visible spectrometric data, confirming the optimal pH level of 7.0 and optimal LigA concentration of $1.5 \mu\text{g mL}^{-1}$ for the functionalisation of silver nanoparticles.

Figure 5 represents the SEM image and UV-visible absorption spectrum of silver nanoparticles. The mean diameter of the synthesised nanoparticles was 60 nm with a surface plasmon resonance peak at 425 nm.

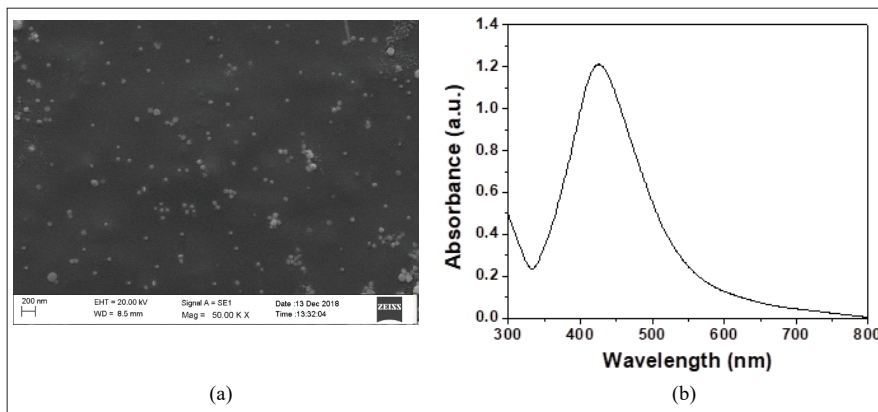


Figure 5: (a) SEM image and (b) UV-visible absorption spectrum of silver nanoparticles

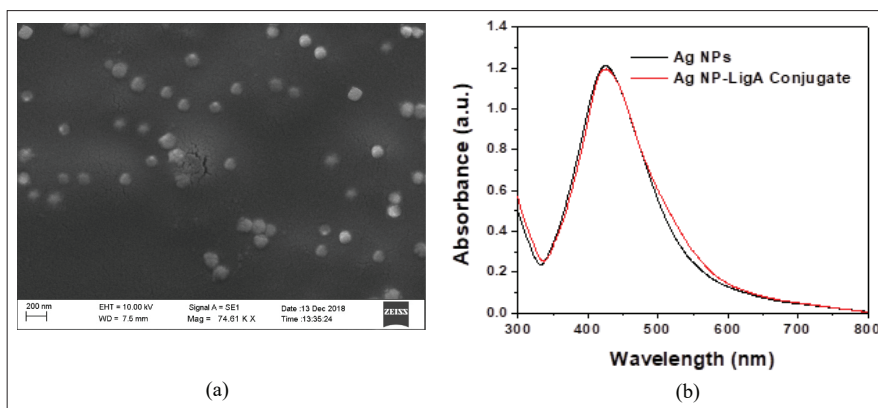


Figure 6: (a) SEM image and (b) UV-visible absorption spectrum of LigA-functionalised silver nanoparticles at pH of 7.0 and 25°C

Figure 6 represents the SEM image and UV-visible absorption spectrum of LigA functionalised silver nanoparticles at optimal environmental conditions (pH 7.0 and 25°C). However, according to the SEM image, the mean diameter of the functionalised nanoparticles was 110 nm. The surface plasmon resonance peak of functionalised nanoparticles was observed at 428 nm.

LigA functionalisation was further identified using the characteristic transmission peaks of the protein in FT-IR spectrum (Figure 7).

Bending vibrations of amide I and amide II bands of the protein were observed at 1642 cm^{-1} and 1540 cm^{-1} respectively, while the bands at 3373 cm^{-1} and 2956 cm^{-1} correspond to the stretching vibrations of amide I and amide II, respectively (Jain *et al.*, 2011). The two bands seen at 1040 cm^{-1} and 1359 cm^{-1} can be assigned to C–N stretching vibrations of aliphatic and aromatic amines, respectively (Vigneshwaran *et al.*, 2006).

Generally, the protein can get directly adsorbed onto nanoparticle surface or the nanoparticle-protein

interaction can take place through cysteine residues or free amine groups or the electrostatic interactions with negatively charged carboxylate groups (Gole *et al.*, 2001). There were no peaks in between 2550 – 2600 cm^{-1} confirming the absence or the presence of an arbitrary amount of S-H bonds, thus cysteine in the protein. In LigA-functionalised nanoparticle sample, enhanced signals of amide I and amide II bands can be observed. Further, IR bands associated with

$-\text{COO}^-$ groups had appeared, including the symmetric/asymmetric stretching vibration of $-\text{COO}^-$ groups which were overlapped with the amide II band at 1564 cm^{-1} , $-\text{COO}^-$ rocking and scissoring at 830 cm^{-1} and 870 cm^{-1} , respectively and $-\text{C}=\text{O}$ out of plane bending at 770 cm^{-1} (Rehman *et al.*, 2010). With the increase in intensities of these bands, it can be assumed that the carboxylic groups are highly exposed at pH 7.0 (Xu *et al.*, 2014).

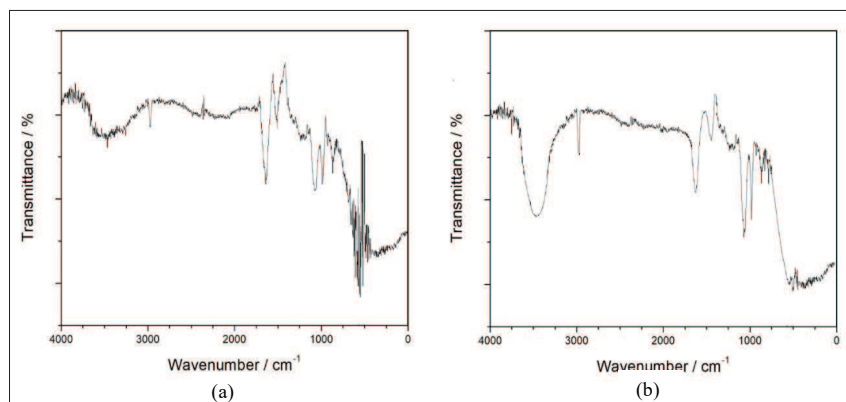


Figure 7: FT-IR spectra of (a) pure LigA protein and (b) LigA-functionalised silver nanoparticles

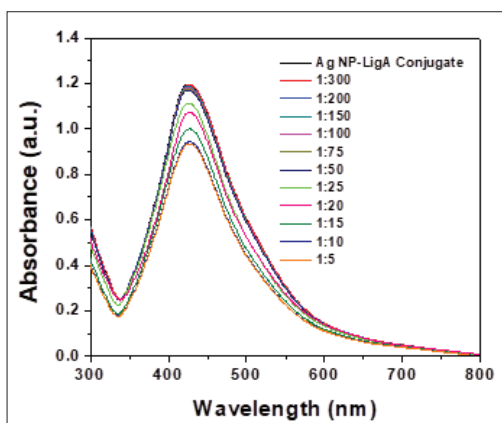


Figure 8: UV-visible absorption spectra of LigA-functionalised silver nanoparticles upon the dilution of anti-leptospiral antibodies from 1:300 to 1:5

It was stated earlier that 110 % of the minimum required amount of protein is appropriate for conjugate stabilisation (Saha *et al.*, 2014). Generally, it is more favourable to use a secondary stabiliser such as bovine serum albumin (BSA) to obtain a higher stability level

of the conjugate in saline environment for a long storage time (Zhang *et al.*, 2014; Taghipour *et al.*, 2018). Therefore, 0.25 % of BSA was added to the colloidal silver conjugate to achieve the highest stabilisation level. There was no significant increase in size indicating that the conjugate was stable in saline environment.

Detection of the sensitivity of LigA-functionalised silver nanoparticles towards anti-leptospiral antibodies

In order to evaluate the sensitivity of LigA functionalised silver nanoparticles, they were tested with anti-leptospiral IgG antibodies. Figure 8 demonstrates the UV-visible absorption upon addition of different amounts of IgG antibody to the nanoparticle conjugate to obtain a dilution series of antibody. The dilution of 1:300 to 1:50 of IgG antibodies maintained no response compared to the nanoparticle conjugate. Subsequently, after 1:25 dilution of IgG antibody, the absorption started to decrease upto 1:10. After that, the system started to saturate, thus, absorption became static. According to these results (Figure 9), sensitivity of the biosensor based on LigA-functionalised silver nanoparticles is acquired to be 1:25 (mean absorbance = 1.113; SD = ± 0.047 ; 95 % CI = 1.100 – 1.126).

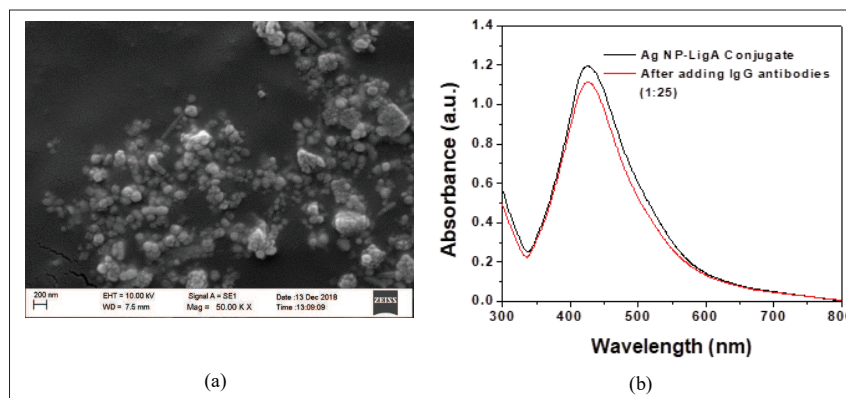


Figure 9: (a) SEM image and (b) UV-visible absorption spectrum of LigA-functionalised silver nanoparticles upon the addition of anti-leptospiral IgG antibodies (1:25 dilution)

Detection of the specificity of LigA-functionalised silver nanoparticles

Furthermore, for the determination of the specificity of the nanoparticle conjugate, they were tested with human serum samples which did not contain anti-leptospiral antibodies. The non-responsive spectra of UV-visible absorption are shown in Figure 10 (mean absorbance = 1.180; SD = ± 0.012 ; 95 % CI = 1.177–1.183). According to these results, it can be confirmed that LigA-functionalised silver nanoparticles were specific to anti-leptospiral antibodies in human sera.

Clinical applications

The LSPR biosensor can be easily utilised to detect IgG antibodies in humans within a shorter screening time. The biosensor needs only a few raw materials to develop and does not require any washing steps or incubation steps in the assay. The results overlap with those of other diagnostic methods, i.e. ELISA or IFA, confirming the accuracy of the method. IgG antibody detection can be performed as a one-step assay, where each batch of samples requires the readings of positive and negative anti-leptospiral IgG serum samples, similar to ELISA, to compare with the test samples. This is the initial step of a promising assay, which can be expanded to detect anti-leptospiral IgM antibodies in recently infected patients of leptospirosis. It is anticipated that this LSPR biosensor can be exploited to detect various proteins by selected nanoparticle-antigen conjugates, which makes the new sensor capable of detecting a wide variety of antibodies. As the next step, the biosensor can be developed to detect and quantify anti-leptospiral IgM antibodies, adhering to the same protocol, which can be used clinically for

acute diagnosis. This is extremely important for endemic countries, such as Sri Lanka, to precisely determine the infection.

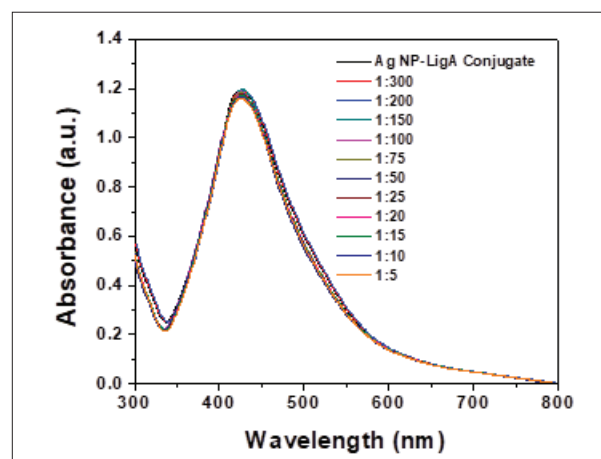


Figure 10: UV-visible absorption spectra of LigA-functionalised silver nanoparticles upon the dilution of anti-leptospiral antibody negative serum samples from 1:300 to 1:5

CONCLUSIONS

In this study, a biosensor based on LigA-functionalised silver nanoparticles was prepared using a straightforward protocol for the detection of anti-leptospiral IgG antibodies in human sera. The main goal of this project was to optimise the essential factors which had a major impact on nanoparticle-antigen conjugation, such as pH, temperature and protein concentration. This LSPR biosensor shows a positive response to anti-leptospiral

IgG protein at pH 7.0 and 25 °C in dilutions lower than 1:25. This novel technique offers substantial improvement in terms of screening time compared to conventional methods. The results can be investigated using UV-visible spectroscopy as a single-step detection method.

This study has been performed with certain limitations. The number of samples anti-leptospirosis IgG antibodies was limited to 50, while the number of samples of anti-leptospirosis IgG antibody negative sera was limited to 20. As the next step, the developed antigen-nanoparticle conjugate can be tested with antibodies of other clinically similar infectious diseases such as Hantavirus, hepatitis and malaria to increase the specificity. Furthermore, the study should report the kinetics of immobilised proteins to identify the rate of antibody-capturing activity of antigens. This can be achieved by analysing small amounts of the sample at regular time intervals for antibodies. Nevertheless, a comprehensive study to analyse the total number of active antigens and the kinetics of antibody capturing is not yet identified.

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